

## X-ray determination of Debye–Waller factors and Debye temperatures of h.c.p. elements Ti, Zr, Ru, Tm, Hf

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Received 3 November 2000  
Accepted 24 November 2000© 2001 International Union of Crystallography  
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Debye–Waller factors of five h.c.p. elements (Ti, Zr, Ru, Tm, Hf) have been determined from X-ray diffraction intensities. Within the limits of experimental errors, the Debye–Waller factors associated with the two principal directions are equal. The Debye temperatures ( $\theta_M$ ) have been evaluated. The energy of formation of vacancies ( $E_f$ ) has been estimated from a semi-empirical relation between  $E_f$  and  $\theta$ .

## 1. Introduction

Gopi Krishna & Sirdeshmukh (1998) recently published a compilation of the Debye–Waller factors and Debye temperatures of h.c.p. elements. From this compilation, it is observed that:

(i) The Debye–Waller factors  $B_c$  and  $B_a$  associated with the  $c$  and  $a$  directions are significantly different only in h.c.p. elements having a  $c/a$  ratio significantly different from the ideal value of 1.63.

(ii) There are no X-ray data on the Debye–Waller factors and Debye temperatures ( $\theta_M$ ) of Zr, Tm and Hf.

(iii) While the Debye temperatures of Ti and Ru have been determined from the temperature variation of X-ray diffraction intensities, there is no explicit attempt to determine the Debye–Waller factors.

The purpose of this communication is to report the results of an X-ray investigation to determine the Debye–Waller factors and Debye temperatures of Ti, Zr, Ru, Tm and Hf.

## 2. Experimental

3N purity powders of grain size < 350  $\mu\text{m}$  were used. Ru and Tm were supplied by Strem Chem, USA, and Ti, Zr and Hf were obtained from the Nuclear Fuel Complex, Hyderabad, India.

Integrated intensities were recorded using a JEOL JDX-8P powder X-ray diffractometer fitted with an NaI (TI) scintillation counter; filtered Cu radiation was employed. The number of reflections recorded ranged from 12 to 16. The procedure for sample preparation, instrumental conditions for recording intensities and the details of the various corrections applied are given in earlier papers (Gopi Krishna *et al.*, 1986; Geeta Krishna *et al.*, 1998).

The procedure for the determination of Debye–Waller factors and  $\theta_M$  from the intensity data and the estimation of associated errors is as discussed by Gopi Krishna *et al.* (1986).

## 3. Results and discussion

The values of the directional Debye–Waller factor ( $B_c$ ,  $B_a$ ) and the mean Debye–Waller factor ( $\bar{B}$ ) for the five h.c.p. elements studied in this work are given in Table 1. It can be seen from the values that for these elements  $B_c \approx B_a$  within limits of the errors. The  $c/a$  values are

Table 1

Debye–Waller factors and Debye temperatures of some h.c.p. elements.

Parameter	$c/a$	$B_c$ ( $\text{\AA}^2$ )	$B_a$ ( $\text{\AA}^2$ )	$\bar{B}$ ( $\text{\AA}^2$ )	$\theta_M$ (K)	$\theta_D$ (K)	$E_f$ (eV)
Reference†	[a]	[b]	[b]	[b]	[b]	[c]	[b]
$\alpha$ -Ti	1.59	0.48 (6)	0.54 (1)	0.52 (3)	391 (22)	420	1.28
Zr	1.59	0.51 (4)	0.52 (1)	0.52 (2)	267 (20)	291	1.37
Ru	1.58	0.15 (1)	0.14 (2)	0.14 (2)	495 (24)	494	3.64
Tm	1.58	0.84 (1)	0.82 (1)	0.83 (1)	155 (10)	167	1.01
Hf	1.58	0.41 (1)	0.41 (2)	0.41 (2)	217 (20)	248	1.72

† Source references: [a] Wyckoff (1960); [b] present work; [c] Gschneidner (1964) for Tm and Singh & Varshni (1982) for the rest.

also given in Table 1; they are also close to 1.63. The near-equality of  $B_c$  and  $B_a$  and the closeness of the  $c/a$  ratio to 1.63 are in conformity with the observation referred to in §1 that the two Debye–Waller factors in h.c.p. elements differ significantly only in those cases where the  $c/a$  ratio differs significantly from 1.63.

The values of  $\theta_M$  are given in Table 1. The values of 391 (22) K and 495 (24) K obtained in this work for the Debye temperatures of Ti and Ru are higher than the values of 270 (30) K and 350 K reported by Spreadborough & Christian (1959) and Brill & Chopra (1962), respectively, from the temperature variation of X-ray intensities. Values of the Debye temperature ( $\theta_D$ ) obtained from specific heat data are also quoted in Table 1. The  $\theta_M$  values agree fairly well with the  $\theta_D$  values although it may be mentioned that exact agreement is not expected.

Glyde (1967) and Tewary (1971) have semi-empirically derived the following relation between the X-ray Debye temperature ( $\theta_M$ ) and energy of formation ( $E_f$ ) of vacancies:

$$E_f = A(k_B/h)^2 M \theta_M^2 a^2,$$

where  $A$  is a constant,  $a$  is the interatomic distance and the other quantities have already been defined. Glyde (1967) showed that for h.c.p. elements the value of  $A$  is  $1.17 \times 10^{-2}$ . Using the values of  $\theta_M$  obtained in this work, values of  $E_f$  have been estimated; these are included in Table 1. Gopi Krishna *et al.* (1986) obtained values in the range 1–1.5 eV for several h.c.p. metals. The  $E_f$  values for the metals studied here are of this order with the exception of Ru, which has a much larger  $E_f$  value.

The referee is thanked for several useful suggestions. The senior author (DBS) wishes to thank the University Grants Commission for financial support during the course of this work. The co-operation of Professor K. G. Subhadra is acknowledged.

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